

**CAUTION**  
**KEEP OUT OF REACH OF CHILDREN**

**Iron deficiencies** appear as yellowing in younger leaves. In severe cases leaves become white but still have green veins. Fruits have poor colour. Iron deficiency often occurs when the soil pH is higher than 7.5 (alkaline soils). Iron sulphate is often used on acid-loving plants, such as camellias and azaleas when the soil is too alkaline and on lawns after heavy periods of rain. Iron Sulphate also helps to lock up phosphorus in the soil, beneficial for some natives. Iron is an essential element for the formation of chlorophyll and helps oxidize sugar for energy.

**MEASURING  
SPOON  
INCLUDED**

- ✓ Soluble & quick acting
- ✓ Corrects yellowing in young leaves
- ✓ Ideal for roses, acid loving plants & lawns

**NET 500g**

**Searles®**

**IRON  
SULPHATE**



**BARCODE**  
**9315138037196**

**DIRECTIONS FOR USE**

**DO NOT SWALLOW**

**EASY GUIDE CHART**

**For soil application only**

**1 HEAPED TEASPOON = 5g**

APPLICATION	(g)	TEA SPOONS	COVERAGE
Ground Application	50g	10 teaspoons	1 square metre

Sprinkle iron sulphate onto soil around plants and water in thoroughly.

**Please Note:** Before correcting iron deficiency in alkaline soils first reduce the pH of the soil by applying Searles Sulphur. Iron sulphate is not effective in alkaline soils.

**IRON SULPHATE ANALYSIS**

30% Iron (Fe) present as Sulphate  
11.5% Sulphur (S) present as Sulphate

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